Formal Syntheses of (-)and (+)-Phalarine

A Journey Start from Oxidative Coupling Ends at Oxidative Coupling



Current Literature Jie Xu 01.29.11

Ding, H.; Chen, D. Angew. Chem. Int. Ed. 2011, 50, 676 – 679

Isolation





Phalaris Coerulescens (Blue Canary Grass)

- In an agronomy-centered investigation directed at the suitability of Phalaris coerulescens for introduction into Australia, Colegate and co-workers isolated a furanobisindole alkaloid.
- Structure confirmed by MS and NMR (not supported by system degradation).
- No promising biological activity has been asserted.

Colegate, S. M. et al. Phytochemistry 1999, 51, 153 – 157.

Biosynthesis



Chan, C.; Li, C.; Zhang, F.; Danishefsky, S. J.; *Tetrahedron Lett.* 2006, 47, 4839 – 4841.

Ring Expansion Strategy





- Migratory aptitude of the urethanebound methylene carbon atom could be rather low.
- H-bonding between the phenolic function and the nitrogen atom could restrict the free rotation.

- Be constructed on-site by ring expansion of an azaspiroindolenine structure
- The rearrangement step would create a positive charge at the C₃, thereby setting the stage for the critical O-C bond formation



Li, C.; Chan, C.; Heimann, A. C.; Danishefsky, S. J. Angew. Chem. Int. Ed. 2007, 46, 1444 – 1447.

Ring Expansion Strategy



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Danishefsky's Racemic Total Synthesis



Rearrangement of azaspiroindolenine to the precursor to phalarine.
Gassman oxindole synthesis while Fischer indolization fails.

Li, C.; Chan, C.; Heimann, A. C.; Danishefsky, S. J. Angew. Chem. Int. Ed. 2007, 46, 1448 – 1450.

Racemic Product



The formation of racemic 18 did not arise from racemization of the substantially enantiopure product.

Retro-Mannich cleavage sequence accounts for the racemization.

Li, C.; Chan, C.; Heimann, A. C.; Danishefsky, S. J. Angew. Chem. Int. Ed. 2007, 46, 1448 – 1450.

Traceless Transfer of Chirality



The chirality inherent in L-tryptophan was transferred to Phalarine in a traceless fashion

Trzupek, J. D.; Lee, D.; Crowley, B. M.; Marathias, V. M.; Danishefsky, S. J. J. Am. Chem. Soc. 2010, 132, 8506

Completion of Enantiopure Synthesis



The absolute configuration was confirmed by X-ray analysis of compound 14.

Trzupek, J. D.; Lee, D.; Crowley, B. M.; Marathias, V. M.; Danishefsky, S. J. J. Am. Chem. Soc. 2010, 132, 8506

Oxidative Coupling Strategy



Muniz K. J. Am. Chem. Soc. **2007**, 149, 1454. Ding, H.; Chen, D. Angew. Chem. Int. Ed. **2011**, 50, 676 – 679.

Formal Synthesis



Ding, H.; Chen, D. Angew. Chem. Int. Ed. 2011, 50, 676 – 679.



Catalyst Screening

Works without palladiumBest with free diol substrate

Pd(OAc)₂ (0.2 equiv), PIDA (1.5 equiv), 1.4:1 1 15 12 CH₂Cl₂, 25°C, 15 min 2 Pd(OAc)₂ (0.2 equiv), CuBr₂ (2.0 equiv), 1:1 15 10 K₂CO₃ (1.1 equiv), CH₂Cl₂, 25 °C, 48 h OMe 15 Pd(OAc)₂ (0.2 equiv), O₂ (1 atm), NaOAc N.D. _ (1.1 equiv), DMF, 25°C, 24 h PIDA (1.5 equiv), CH₂Cl₂, 25 °C, 15 min Δ 15 30 1.4:15 15 PIFA (1.5 equiv), CH₂Cl₂, 25 °C, 15 min 40 1.4:1 6 PIFA (1.5 equiv), CH₃CN, 25°C, 10 min 15 45 1.4:1 7 15 PIFA (1.5 equiv), CH₃CN/H₂O (20:1), 42 1.4:1 25°C, 10 min 8 PIFA (1.5 equiv), CF₃CH₂OH, 25°C, 15 N.D. _ 15 min 9 15 PIFA (1.5 equiv), THF, 25 °C, 15 min 21 1:1 10 15 PIFA (1.5 equiv), toluene, 25°C, 15 min 40 1:2 11 15 PIFA (1.5 equiv), toluene, $-5 \rightarrow 0^{\circ}$ C, 51 1:1.3 35 min 12 PIFA (1.5 equiv), CH₃CN, $-5 \rightarrow 0^{\circ}$ C, 54 15 2:1 30 min 13 DDQ (10 equiv), THF, 25 → 70 °C, 48 h 21 15 anti only 14 15a PIFA (1.5 equiv), CH₃CN, $-5 \rightarrow 0^{\circ}$ C, 42 1.1:1 30 min 15b PIFA (1. equiv), CH_3CN , $-5 \rightarrow 0^{\circ}C$, 15 60 1.4:1 30 min 16 PIFA (1.1 equiv), CH₃CN, $-5 \rightarrow 0^{\circ}$ C, 3:1 2 46 15 min 4 17 2 PIFA (1.2 equiv), CH_2Cl_2 , $-5 \rightarrow 0$ °C, 68 9:1 30 min

Entry Sub. Conditions

[a] Yields refer to chromatographically and spectroscopically homogeneous materials. [b] Determined by ¹H NMR spectroscopic analysis of the crude reaction mixture. The ratios are approximate. DDQ = 2,4-dichloro-5,6-dicyanobezoquinone, N.D. = not detected.

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Product syn/

anti^[b]

Yield

[%]^[a]



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Summary





Realization of oxidative coupling

N-CO₂Me

OMe

4

н

MeO.

PIFA

CH₃CN



OMe

ÔH

OMe

3

N-CO₂Me

2